isolation très-considérable. Il est aussi facile de voir que la quantité de l'électricité développée par cette excitation continuelle qui a lieu dans les circuits doit être d'autant plus grande que le circuit est plus parfait conducteur. Ainsi, le circuit thermo-électrique donnera une quantité d'électricité incomparablement plus grande que celle qu'on pouvait tirer d'aucun autre appareil qu'on ait inventé jusqu'à notre temps. Si à l'aide des anciens circuits on a décomposé successivement l'eau, les acides et les alcalis, il n'est pas hors des limites de la vraisemblance qu'on parviendra, par le nouveau, à décomposer les métaux mêmes, et à compléter ainsi le grand changement qu'a commencé dans la chimie la pile de Volta.

## ON AN APPARENT PARADOXICAL GALVANIC EXPERIMENT

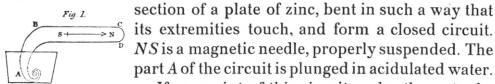
## BY J. C. OERSTED

PROFESSOR OF CHEMISTRY AND NATURAL PHILOSOPHY IN THE UNIVERSITY OF COPENHAGEN, AND F. R. S. E. COMMUNICATED BY THE AUTHOR

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In a Memoir, published some months ago, by M. Von Moll,<sup>2</sup> at Utrecht,<sup>3</sup> this philosopher (already known from various experimental researches) describes an experiment, which, at first sight, appears to indicate a new class of galvanic phenomena.

I have submitted this experiment to an attentive examination. Fig. 1. is the apparatus of M. Von Moll. ABCD is a perpendicular



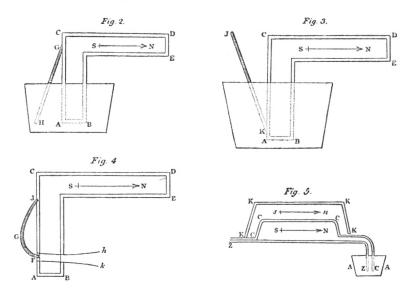
If any point of this circuit under the water be touched by a piece of brass, the motion of the needle indicates an electric current. In order to be certain that the metallic continuity was not interrupted by the interposition of a part of the fluid, I

<sup>&</sup>lt;sup>1</sup> [Among H. C. Oersted's papers a French rendering of the same essay has been found (parcel 23). No printed version has been discoverable.]

<sup>&</sup>lt;sup>2</sup> [o: Moll]. <sup>8</sup> Edinburgh Phil. Journal, vol. 1x. p. 167.

substituted for that in Fig. 1., the circuit ABCDE, Fig. 2., cut out of a plate of zinc. The effect described by the Dutch philosopher was produced by this circuit likewise; but I soon discovered that it was owing to the ordinary galvanic circuits, like that formed by the copper GH, the zinc GA, and the fluid between H and A, or, as in that in Fig. 3. formed by the copper JK, the zinc KC, and the fluid.

The contact of the copper and zinc above the water, or at the surface of the water, produces no effect. In order to make myself



sure that a collateral galvanic circuit was capable of producing such an effect on a homogeneous metallic circuit, I made the construction shewn at Fig. 4., in which ABCDE is the same homogeneous circuit as before, but JGF is an arch of copper, in contact with the zinc at J, and separated from it at F by a fold hk of paper wetted in acidulated water. In alternately opening and closing this circuit, I found that the needle moved as in the preceding cases. This construction may therefore be considered as composed of a galvanic circuit JGFJ, and a conductor JCDEBAF, which transmits a current similar to that transmitted by JGF.

This explanation is confirmed by an experiment made with the construction represented in Fig. 5. in which ZZ is a plate of zinc, CCCC a plate of copper, and AA a vessel of acidulated water. When a magnetic needle is placed at NS, it is deflected according to the known laws; but, if to this circuit there be added the conductor KKK, a part of the electricity passes in it, and acts more

feebly on the needle, from being at a greater distance from it. The effect of the second conductor becomes more striking when the needle is placed at ns, and when (after having noted its deviation) the conductor KKK is added; because, in this case, the second conductor being above the needle, tends to give it a contrary deflexion to that given by the first conductor, which is below it. These experiments have the same result when ZZ is made of copper, and CCC of zinc.

On applying all this to the constructions in Figs. 2. and 3., we observe, that CDEBA is the same thing as the second conductor in Fig. 5. and that the current in the part DE (Figs. 2. and 3.) should have the same direction as in the part CA (to which it is parallel), in the same manner that the currents are similar in CCC and KKK, in Fig. 5. This being granted, we can determine the direction of the current in all the other parts of CDEBA (Figs. 2. and 3.), and experiments with the needle will confirm the predictions of the theory. In some experiments, M. Von Moll substituted a plate of zinc for the copper, with which he touched the zinc circuit, and produced electro-magnetic effects by these means also. This is likewise reducible to a collateral galvanic circuit; for I have proved, by experiments published two years ago, that a galvanic circuit may be made for a short space, by two plates of zinc and a liquid, provided that one of the plates be brought into contact with the liquid before the other.